Tel: 0086-571-56623320 Fax:0086-571-56623318

E-mail:sales@sunlongbiotech.com

www.sunlongbiotech.com

# Dehydroascorbic Acid(DHA)ContentAssay Kit

Note: Take two or three different samples for prediction before test.

**Detection equipment:** Ultraviolet spectrophotometer

Cat No: AK0466 **Size:**50T/48S

# **Components:**

Extract solution:Liquid 60 mL×1, store at room temperature.

Reagent I: Liquid40 mL×1, store at room temperature.

Reagent II: Powder×1, store at -20°Cand avoid light; Add 5 mL ofdistilled water before use, mix thoroughly. Store at -20°C after spacing out.

Standard: Powder×1, store at -20°C and avoid light. Add 5.743 mL of distilled water to dissolve, then take 0. 1mL and putitin 0.9 mL of distilled water, mix thoroughly and to be prepared as 0.1 µmol/mL DHA. Store at -20°C after dividing the solution into several parts.

# **Description:**

AsA is an important indicator of plant cells, the content of AsA, redox state(AsA/DHA) and its synthesis and metabolism-related enzyme activities are related to the response of plants to a series of environmental stress. DHA is a reversible oxidized form of AsA. It forms a redox system with ascorbic acid in the living body and has the function of an electron acceptor.

DTT deoxidize DHA to form AsA. According to the generation rate of AsA, can calculate the content of DHA.

### **Technical Specifications**

Minimum Detection Limit: 0.0065 µmol/mL

Linear Range: 0.015625-1 μmol/mL

# Required but not provided

Low temperature centrifuge, ultraviolet spectrophotometer, 1 mL quartz cuvette, adjustable pipette, mortar/homogenizer, ice and distilled water.

### **Protocol:**

# I. DHA Extraction:

Add 1 mL extract solution to 0.1 g of sample, fully grind on ice. centrifuge at 16000 g and 4°C for 20 min. Supernatant is ready for test.

### II. Procedure

- 1. Preheat ultraviolet spectrophotometer for 30 min, adjust wavelength to 265 nm, set zero with distilled water.
- 2. Preheat Reagent I at 25°C water bath for 30 min.





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- 3. Standard tube: Add 100  $\mu$ L of standard, 800  $\mu$ L of Reagent I and 100  $\mu$ L of Reagent II to 1 mL quartz cuvette, mix thoroughly and quickly, detect at 265 nm, record the absorbance at 10s and 130s. Record A1,A2, $\Delta$ A<sub>S</sub>=A2-A1.
- 4. Test tube: Add 100  $\mu$ L of supernatant, 800  $\mu$ L of Reagent I and 100  $\mu$ L of Reagent II to 1 mL quartz cuvette, mix thoroughly and quickly, detect at 265 nm, record the absorbance at 10s and 130s.Record A3,A4, $\Delta$ A<sub>T</sub>=A4-A3.

### III. Calculation

1. Protein concentration

DHA( $\mu$ mol/mg prot) =C<sub>S</sub>×( $\Delta$ A<sub>T</sub>÷ $\Delta$ A<sub>S</sub>)÷Cpr=0. 1× $\Delta$ A<sub>T</sub>÷ $\Delta$ A<sub>S</sub>÷Cpr

2. Sample weight

DHA( $\mu$ mol/g) = [ $C_S \times \Delta A_T \div \Delta A_S \times V_{ST}$ ]  $\div W = 0.1 \times \Delta A_T \div \Delta A_S \div W$ 

C<sub>S</sub>: DHA concentration, 0.1 μmol/mL;

V<sub>ST</sub>: Supernatant total volume, 1.0 mL;

Cpr: Supernatant protein concentration, mg/mL;

W: Sample weight, g.

#### Note:

Before the formal experiment, do  $1\sim2$  pre-experiments to ensure that the value of  $\Delta A$  is less than 0.4.

# **Experimental instances:**

1. Take 0. 1g of photinia, add 1mL of extract reagent, homogenate on ice. Centrifuge at 16000g for 20 minutes at 4 °C , take the supernatant, dilute four times, and test according to the measured steps. Calculate  $\Delta A_T = A4 - A3 = 1.207 - 1.19 = 0.017$ ,  $\Delta A_S = A2 - A1 = 0.295 - 0.043 = 0.252$ , calculate the enzyme activity according to sample weight:

DHA( $\mu$ mol/g weight) =0.5 ×  $\Delta$  A<sub>T</sub> ÷  $\Delta$  A<sub>S</sub> ÷ W×4 (dilution ratio) =1.349  $\mu$ mol/g weight.

### **Related products:**

AK0464/AK0463 L-galactose- 1,4-lactone dehydrogenase (Gal LDH)Assay Kit

AK0462/AK0461 Ascorbic Acid Oxidase(AAO)Activity Assay Kit AK0576/AK0575 Ascorbate Peroxidase (APX) Activity Assay Kit